DITHIOESTERS AS HETERO-DIENOPHILES IN DIELS-ALDER CYCLOADDITION -SYNTHESIS OF 6-ALKYLTHIO-5,6-DIHYDRO-2H-THIOPYRANS.

Pierre BESLIN and Patrick METZNER

Laboratoire de Chimie des Composés Thioorganiques, associé au CNRS, ISMRA, Université 14032 CAEN (France).

The Diels-Alder reaction of methyl alkanedithiocarboxylates with various dienes at 160 °C affords 6-alkylthio-5,6-dihydro-2<u>H</u>-thiopyrans. The regioselectivity and the scope of the reaction were examined.

Dithioesters bearing a strong electron-withdrawing group, such as CF_3CS_2Et and NC-CS₂Me, are very reactive heterodienophiles.¹ The Diels-Alder reaction with butadienes proceeds respectively at 0° and -78°. In the literature,² there is no example of this cycloaddition involving dithioesters which do not bear an electron-withdrawing group. Our interest in thiocarbonyl chemistry led us to explore this reaction with the now easily available alkanedithiocarboxylates.³ The expected dihydrothiopyrans could be versatile intermediates ⁴ in the synthesis of functionalyzed Z-olefins with high isomeric purity.

As substrates we selected methyl dithioacetate $\underline{1}$, dithiopropanoate $\underline{2}$, dithioisobutyrate $\underline{3}$ and dithiobenzoate $\underline{4}$. A solution of <u>ca</u> 10 mmol dithioester, 5 fold excess diene and a catalytic amount of hydroquinone in benzene or toluene was heated in a vacuum sealed tube for a standard period of 3 days at 160°. The cooled mixture is concentrated <u>in vacuo</u>, analyzed by NMR, purified by medium pressure liquid chromatography and bulb to bulb distillation.⁵

With symmetrical dienes (butadiene and 2,3-dimethylbutadiene) and dithioesters 1, 2, good yields of products 5, 6, 7 are obtained. Dithiobenzoate also reacts well in

	s +	$\sum_{R^2}^{R^2}$	160° 3 days	$\overset{S}{\underset{MeS}{\overset{R^{1}}{\overset{R^{2}}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{R}}{\overset{R^{2}}{\overset{R^{R}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R^{2}}{\overset{R}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}}$
	R ¹ SMe		PhH or PhMe	
1	$R^{1} = Me$	R²≃ H		<u>5</u> 60 %
		R²= Me		<u>6</u> 73%
2	R ¹ = Et	R²= H		<u>7</u> 50 %
3	$R^{1} = isoPr$	R²= H		8 20 %
<u>4</u>	$R^1 = Ph$	R ² = Me		9 (70%)

these conditions but attempts to isolate the product $\underline{9}$ led to substantial decomposition. In the case of dithioisobutyrate 3, a poor yield of $\underline{8}$ is obtained, probably in relation with its steric hindrance and its propensity towards enethiolization.

We then focused our attention on the regioselectivity of the reaction of methyl dithioacetate 1 with unsymmetrical dienes : isoprene and 1,3-pentadiene. With isoprene, a 1 : 1 mixture (77 % yield) of two unseparated isomers 10 and 11 is obtained. Reaction with trans-1,3-pentadiene exhibits a slight regioselectivity 2 : 1 in favour of 2-substituted adducts. GC separation of the mixture (75 % yield) affords pure 2-Me isomers (ratio 12 : 34%; 13 : 29\%) and a 1 : 1 mixture of the 5-Me isomers (14 + 15 : 37%).



Thus, Diels-Alder addition of dithioesters, which do not possess an electrowithdrawing substituant at the thiocarbonyl site, with dienes is feasible and furnishes reasonable yields of 6-alkylthio-5,6-dihydro-2H-thiopyrans in the case of R= Me, Et.

In order to compare the reactivity of various thiocarbonyl compounds as heterodienophiles, we also treated methyl trithiocarbonate with butadiene under the same conditions and noted that the starting material was recovered. From these results and literature reports, the following reactivity order is suggested :



It is also noteworthy that the preceding reactions occur under much milder conditions than with oxygenated analogues.¹⁰

We are currently investigating the possibility of improving the regioselectivity via a catalytic effet, and the utilization of dihydrothiopyrans in the synthesis of natural Z-olefins.

References and notes

- (a) W.J. Middleton, <u>J.Org.Chem</u>, <u>30</u>, 1390 (1965); (b) D.M. Vyas and G.W. Hay, 1 J.C.S. Perkin I, 180 (1975) and cited preceding references.
- 2 S. Scheithauer and R. Mayer, "Thio-and dithiocarboxylic acids and their derivatives" in Topics in Sulfur Chemistry Vol. 4, Ed. A. Senning, Georg Thieme, Stuttgart,
- 1979, 295. J.M. Beiner and A. Thuillier, <u>C.R.Acad.Sci.</u>, <u>274</u>, C, 642 (1972); J. Meijer, P. Vermeer and L. Brandsma, <u>Rec. Trav. Chim.</u>, <u>92</u>, 601 (1973). 3
- 4 P.L. Stotter and R.E. Hornish, J.Amer.Chem.Soc., 95, 4444 (1973); S. Torii, H. Tanaka and Y. Tomotaki, <u>Bull.Chem.Soc.Japan</u>, 50, 537 (1977). 5 Satisfactory spectral data (¹H, ¹³C NMR, IR and Mass) and microanalyses were
- obtained for all new compounds.
- 6
- D.R. Dice and R.P. Steer, <u>Canad.J.Chem.</u>, 52, 3518, (1974). A. Ohno, Y. Ohnishi and G. Tsuchihashi, Tetrahedron, 25, 871 (1969); J.F. Biellmann, 7 J.B. Ducep and J.J. Vicens, Ibid, 32, 1801 (1976). H. Wollweber, "Diels-Alder Reaktion", Georg Thieme Verlag, Stuttgart, 1972, 199.
- 8 0 This work.
- 10 For example see ref (8) p. 196. For a recently published paper, see T. Kobayashi and H. Tsuruta, Synthesis, 492 (1980).

(Received in France 31 July 1980)